


Non-equilibrium thermodynamics

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Outline

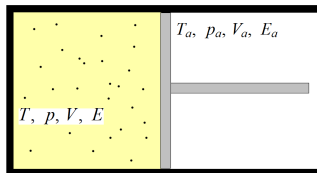
- 1 Introduction - Many faces of thermodynamics
- 2 Thermostatistics
- 3 Ordinary thermodynamics
 - Ordinary thermomechanics with inertia
- 4 Heat conduction
 - Heat conduction beyond Fourier
- 5 Rheology of solids
- 6 Summary

Introduction

Thermodynamics		
Equilibrium	$T, E, T(E)$	statics
Ordinary	$E(t), T(E(t))$	homogeneous
Non-equilibrium	$e(t, x^i), T(e(t, x^i))$	field theory

Field theories: relative/Galilean relativistic/special relativistic/general relativistic

Simple system :



Thermostatistics - potentials

Extensivity

Existence of densities and specific quantities. Transition to local, field quantities.

Gibbs relation

$$dE = TdS - pdV + \mu dM, \quad dM = 0.$$

Meaning: Entropy, $S(E, V)$, is a potential.

$$\partial_E S = \frac{1}{T}(E, V), \quad \partial_V S = \frac{p}{T}(E, V)$$

Vector field, integrating multiplier.

Example: ideal gas

$$E = cMT, \quad pV = MRT$$

Thermostatistics - extensivity

Bulk relations

$$dE = TdS - pdV + \mu dM, \quad E = TS - pV + \mu M.$$

Specific quantities: $E = Me, S = Ms, V = Mv$

$$\mu := e - Ts + pv, \quad \text{definition!}$$

Specific relations

$$de = Tds - pdv, \quad \mu := e - Ts + pv$$

Density relations

$$d\rho_E = Td\rho_S + \mu d\rho, \quad \rho_e + p = \rho_h = T\rho_s + \mu\rho$$

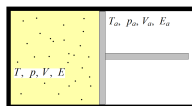
Ordinary thermodynamics

Processes:

$$\dot{E} = (cT(t))' = -\alpha(T(t) - T_a), \quad T(0) = T_i.$$
$$T(t) = T_a + (T_i - T_a)e^{-\frac{\alpha}{c}t}$$

System paradox

Body and environment



$$\dot{E} = \hat{Q} + \hat{W} \quad ? \quad \dot{E} = T\dot{S} - p\dot{V}$$

Evolution equation:

$$\dot{E} = \hat{Q} + \hat{W} \quad \dot{V} = \hat{F}.$$

Stability and second law

Second law:

- $S(E, V)$ is potential,
- $S(E, V)$ is concave (thermodynamic stability),
- $S(E, V)$ is increasing (?!, insulated).

Similarity:

Ljapunov theorem:

$\dot{x} = f(x)$, $f(x_0) = 0$ autonomous d.e., x_0 equilibrium.

- There is a Ljapunov function $L(x)$, of x_0 of the d.e.
- L has a strict maximum at equilibrium ($DL_0 = 0$, D^2L_0 is concave),
- L is monotonously increasing along the d.e., DL has a strict max.

Intensives and equilibrium.

Evolution equation

Equation of 'motion':

$$\dot{E} = \hat{Q} - p\hat{F}, \quad \dot{V} = \hat{F};$$

classical work, volume evolution, \hat{Q} , \hat{F} are constitutive

Equilibrium: $T = T_0$, $p = p_0$, $(\hat{Q}, \hat{F})(T, T_0, p, p_0)$ (derived).

Entropy? But: $\dot{S} = \dots = \frac{\hat{Q}}{T}$

Ljapunov function:

$$L(E, V) = S(E, V) - \frac{1}{T_0}E - \frac{p_0}{T_0}V$$

Equilibrium! $DL(E, V) = \left(\frac{1}{T} - \frac{1}{T_0}, \frac{p}{T} - \frac{p_0}{T_0}\right)$

Conditions of increasing:

$$\begin{aligned} \dot{L}(E, V) &= DL \cdot (\dot{E}, \dot{V}) = \left(\frac{1}{T} - \frac{1}{T_0}, \frac{p}{T} - \frac{p_0}{T_0}\right) \cdot (\hat{Q} - p\hat{F}, \hat{F}) = \dots = \\ &= \left(\frac{1}{T} - \frac{1}{T_0}\right) \hat{Q} + \frac{1}{T_0}(p - p_0)\hat{F} \geq 0. \end{aligned}$$

It is simple and reasonable.

Second Law

Requirement of stability:

$$\dot{E} = \hat{Q} - p\hat{F}, \quad \dot{V} = \hat{F};$$

$$\left[L(E, V) = S(E, V) - \frac{1}{T_0}E - \frac{p_0}{T_0}V \right]$$

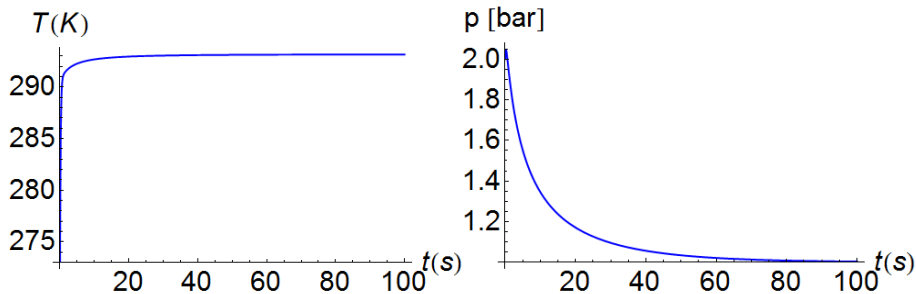
$$\left(\frac{1}{T} - \frac{1}{T_0} \right) \hat{Q} + \frac{1}{T_0} (p - p_0) \hat{F} \geq 0.$$

Linear relations, Onsager.

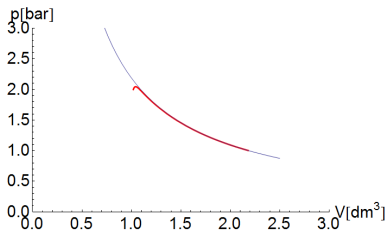
What is L ?

- Entropy of a reservoir.
- Exergy.

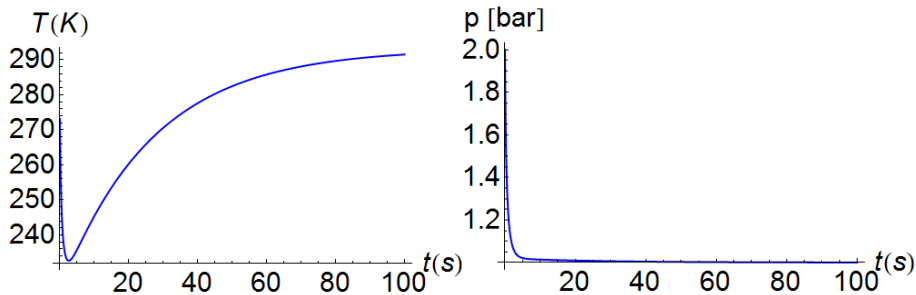
Example 1.1



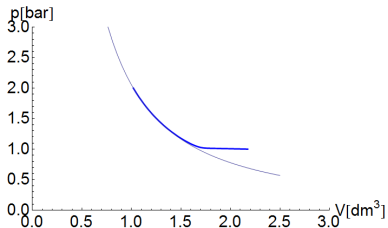
Fast thermal and slow mechanical equilibration



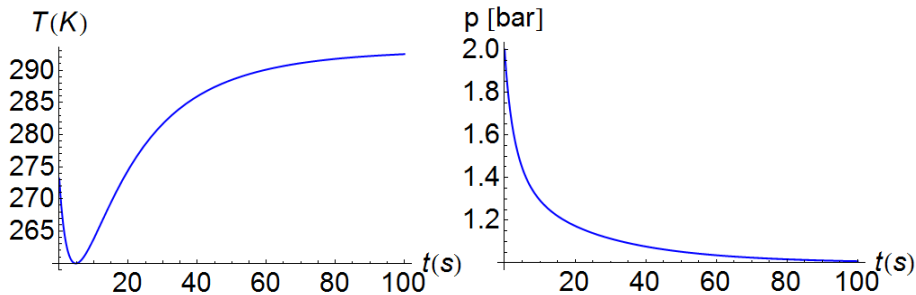
Example 1.2



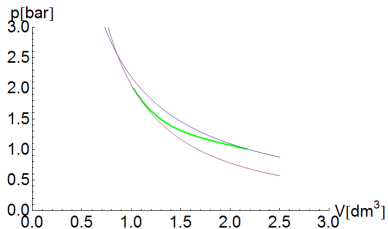
Fast mechanical equilibration: adiabatic process followed by an isobaric one



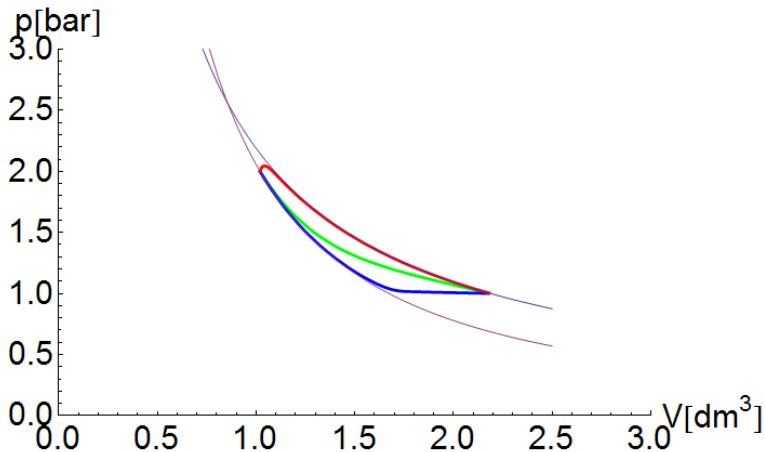
Example 1.3



Intermediate speed.



Example 1.4



The isothermal (red), adiabatic-isobaric blue and mixed (green) processes together with the $T = T_a$ isotherm and $p = p_i(V/V_i)^\kappa$ adiabatic curves (thin lines) on the p-V plane.

Ordinary thermomechanics I.

Equation of 'motion':

$$\dot{E} = \hat{Q} - p\hat{F} + MU\hat{G}, \quad \dot{V} = \hat{F}, \quad \dot{U} = \hat{G}.$$

total energy $E_T = E + M\frac{U^2}{2}$; \hat{Q} , \hat{F} , \hat{G} are constitutive

Equilibrium: $T = T_0$, $p = p_0$, $U = 0$, $(\hat{Q}, \hat{F}, \hat{G})(T, T_0, p, p_0, U)$ (derived).

Ljapunov function:

$$L(E, V) = S\left(E_T - M\frac{U^2}{2}, V\right) - \frac{1}{T_0}E - \frac{p_0}{T_0}V$$

Equilibrium! $DL(E, V, U) = \left(\frac{1}{T} - \frac{1}{T_0}, \frac{p}{T} - \frac{p_0}{T_0}, -\frac{MU}{T}\right)$

$$\begin{aligned} \text{Conditions of increasing: } \dot{L}(E, V, U) &= DL \cdot (\dot{E}, \dot{V}, \dot{U}) = \\ &= \left(\frac{1}{T} - \frac{1}{T_0}, \frac{p}{T} - \frac{p_0}{T_0}, -\frac{MU}{T}\right) \cdot (\hat{Q} - p\hat{F} + mU\hat{G}, \hat{F}, \hat{G}) = \dots = \\ &= \left(\frac{1}{T} - \frac{1}{T_0}\right) \hat{Q} + \frac{1}{T_0}(p - p_0)\hat{F} - \frac{MU}{T_0} \hat{G} \geq 0. \end{aligned}$$

Ordinary thermomechanics II.

Equation of 'motion':

$$\dot{E} = \hat{Q} - p\hat{F} + MU\hat{G}, \quad \dot{V} = \hat{F}, \quad \dot{U} = \hat{G}.$$

Linear, simplified constitutive equations:

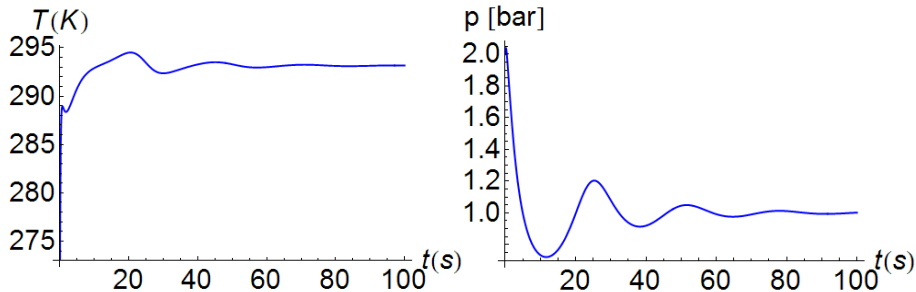
$$\begin{aligned}\hat{Q} &= -\alpha(T - T_0), \\ \hat{F} &= \beta_1(p - p_0) - b_{12}MU, \\ \hat{G} &= \beta_{21}(p - p_0) - b_2MU.\end{aligned}$$

Pure mechanics: $\alpha = 0$

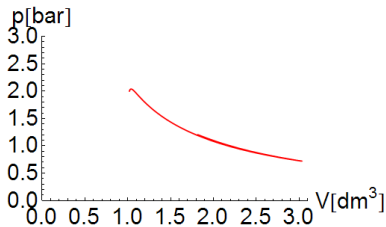
Usual dissipative mechanics: $\alpha = 0, \beta_{12} = -\beta_{21}, \beta_1 = 0$

Usual ideal mechanics: $\alpha = 0, \beta_{12} = -\beta_{21}, \beta_1 = \beta_2 = 0$

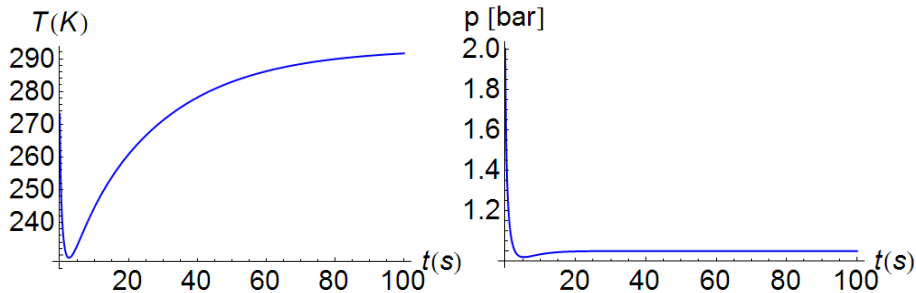
Example 2.1



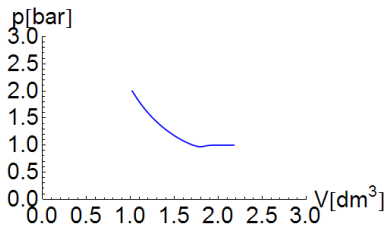
Fast thermal and slow mechanical equilibration: isothermal oscillations



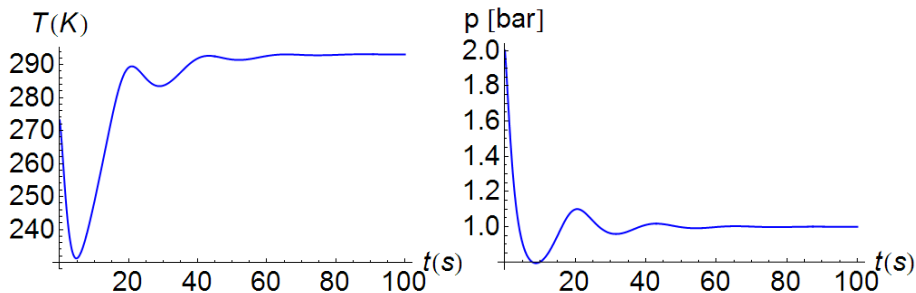
Example 2.2



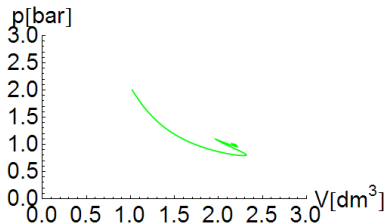
Fast mechanical equilibration: adiabatic process followed by an isobaric one: no oscillations



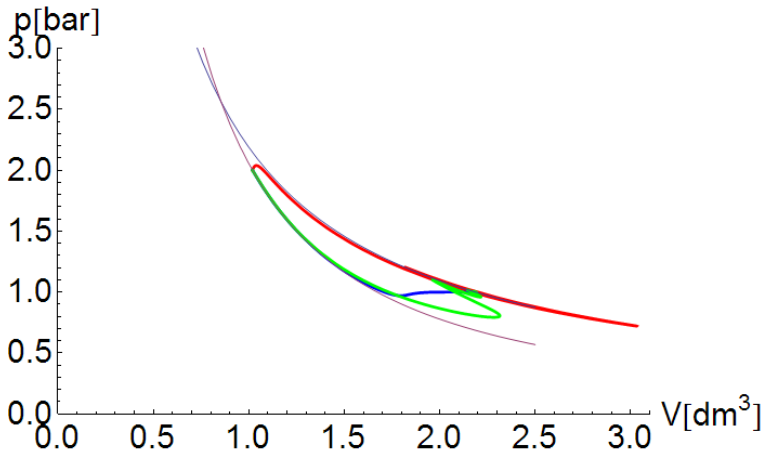
Example 2.3



Intermediate speed, with oscillations.



Example 2.4



The isothermal (red), adiabatic-isobaric blue and mixed (green) processes together with the $T = T_a$ isotherm and $p = p_i(V/V_i)^\kappa$ adiabatic curves (thin lines) on the p-V plane. Oscillations galore.

Heat conduction

Heat conduction theory - Fourier

Balance of internal energy

$$\rho \dot{e} + \partial_i q^i = 0$$

ρ density, e specific internal energy, q^i conductive current density of the internal energy, heat flux, rigid conductors: $\dot{} = \partial_t$, $\rho = \text{const.}$

Thermodynamics: $s(e)$, $de = Tds$

Entropy balance

$$\rho \dot{s} + \partial_i J^i \geq 0$$

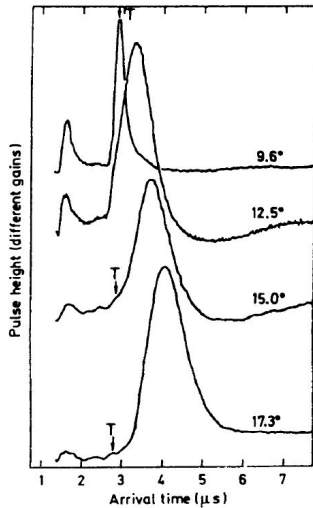
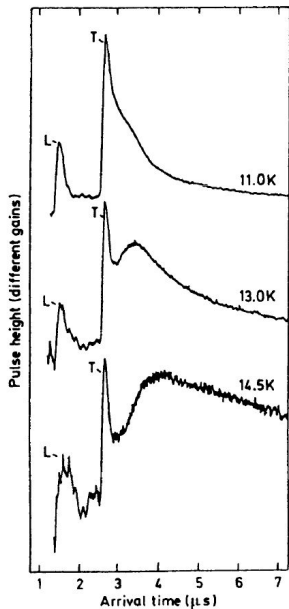
s specific entropy, J^i entropy flux. Assumption: $J^i = \frac{q^i}{T}$

Entropy production

$$\rho \dot{s} e + \partial_i \frac{q^i}{T} = \dots = q^i \partial_i \frac{1}{T} \geq 0$$

Fourier law: $q^i = \lambda \partial_i \frac{1}{T} = -\lambda_F \partial_i T$

Heat pulse experiments: low temperature NaF

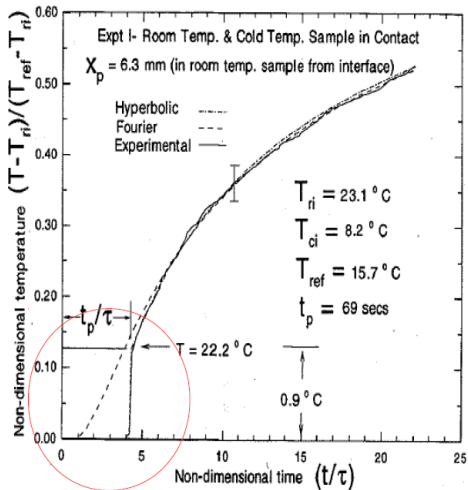
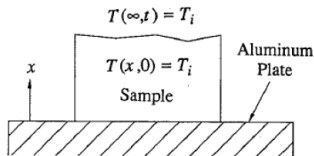


Frozen meat experiment - original

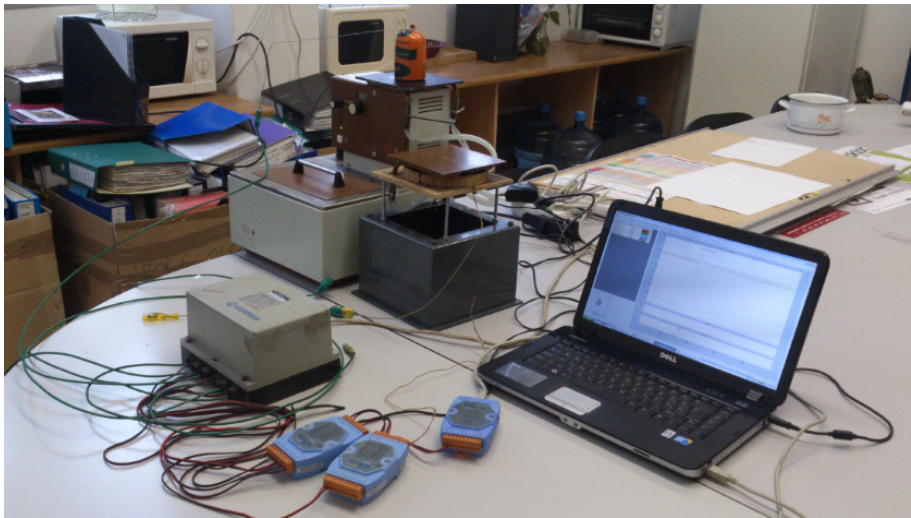
Mitra-Kumar-Vedavarz-Moallemi, 1995

Processed frozen meat:

$$\tau = 20-60 \text{ s}$$



Frozen meat experiment - home version 1



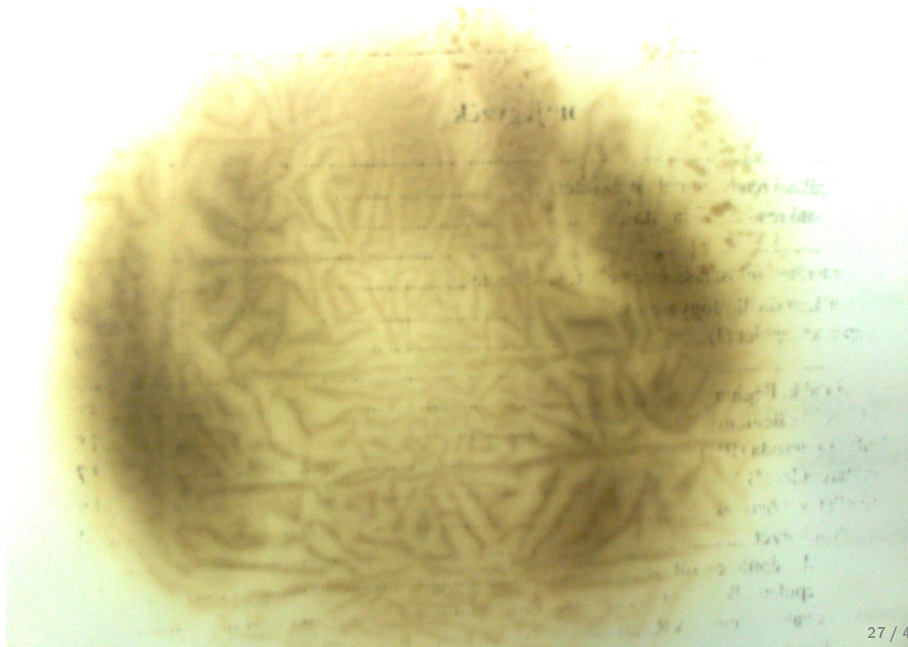
Frozen meat experiment - home version 2



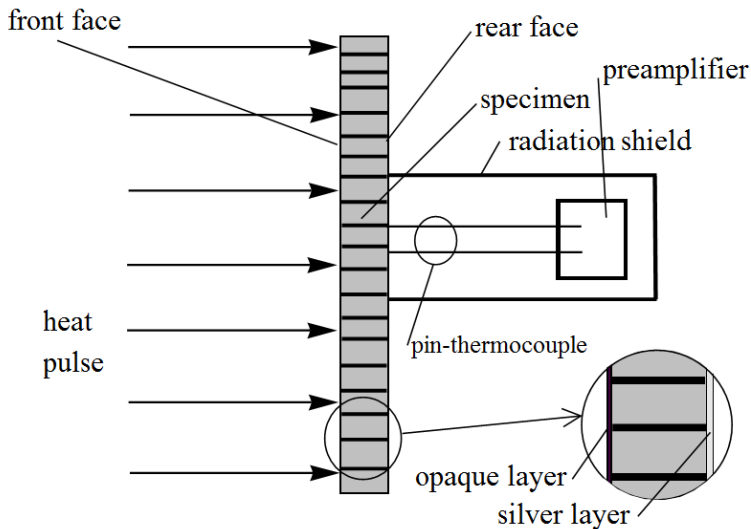
Roasted book experiment



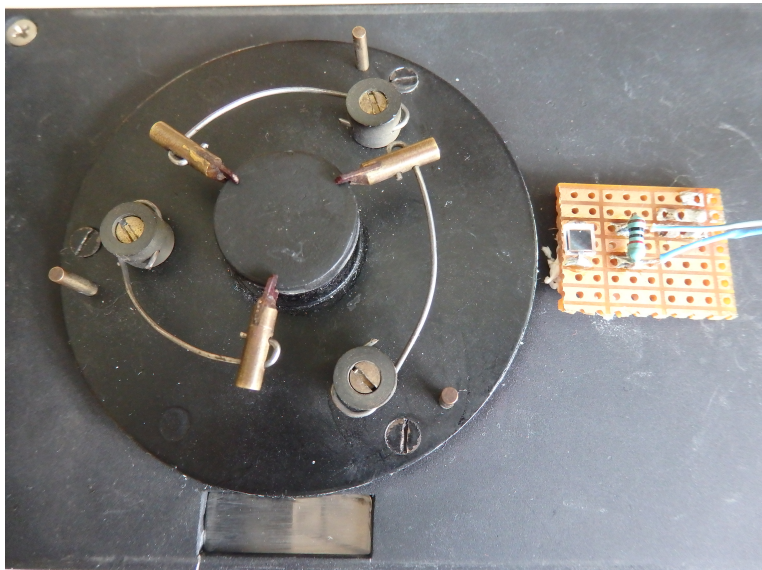
Roasted book experiment - the result



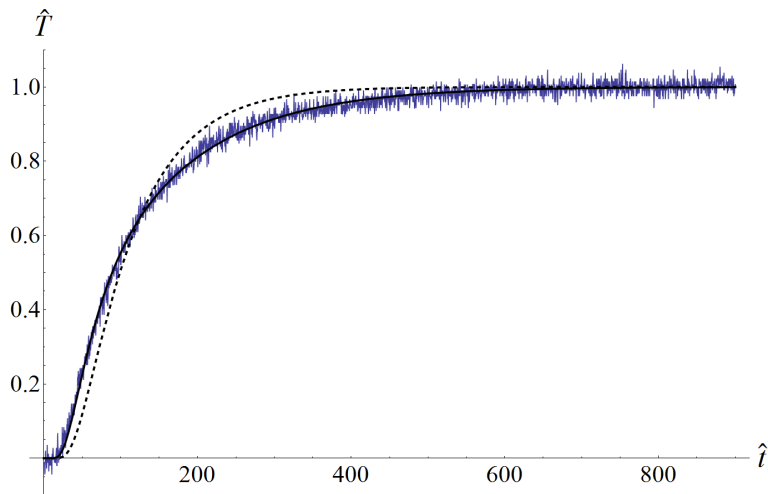
Heat pulse experiment - sketch



Heat pulse experiment - device



Heat pulse experiment - result



Metal-plastic layers, Fourier and GK fits.

Heat conduction theory - beyond Fourier

Balance of internal energy

$$\rho \dot{e} + \partial_i q^i = 0$$

Beyond Fourier: *memory* and *nonlocal* extensions

- ① Fourier: $q = \lambda_1 \partial_x T$
- ② Maxwell-Cattaneo-Vernotte: $\tau_q \partial_t q + q = \lambda_1 \partial_x T$
- ③ Jeffreys type (or dual-phase lag): $\tau_q \partial_t q + q = \lambda_1 \partial_x T + \lambda_2 \partial_{tx} T$
- ④ Guyer-Krumhansl: $\tau_q \partial_t q + q = \lambda_1 \partial_x T + a \partial_{xx} q$
- ⑤ Green-Naghdi: $\tau_q \partial_t q = \lambda_1 \partial_x T + a \partial_{xx} q$
- ⑥ Cahn-Hilliard type: $q = \lambda_1 \partial_x T + a \partial_{xx} q$
- ⑦ There are more ...

Heat conduction beyond Fourier

Internal variable + generalized entropy current:

- ξ^i is an *internal variable*
- entropy density: $s(e, \xi^i) = s_e \left(e - \frac{m}{2} \xi^i \cdot \xi^i \right)$
- entropy current: $J^i = B^{ij} q^j$

Entropy production:

$$\left(B^{ij} - \frac{1}{T} \delta^{ij} \right) \partial_i q_j + q_j \partial_i B^{ij} - \rho \frac{m}{T} \xi_i \dot{\xi}^i \geq 0$$

Isotropic, linear relations between *thermodynamic fluxes* and *forces*:

$$\begin{aligned} B^{ij} - \frac{1}{T} \delta^{ij} &= k_1 \partial_k q^k \delta^{ij} + k_2 \frac{1}{2} (\partial^i q^j + \partial^j q^i) + k_3 \frac{1}{2} (\partial^i q^j - \partial^j q^i) \\ q^j &= l_1 \partial_j B^{ji} - l_{12} \rho \frac{m}{T} \xi^i \\ \dot{\xi}^i &= l_{21} \partial_j B^{ji} - l_2 \rho \frac{m}{T} \xi^i \end{aligned}$$

Simplification in 1+1D

The system :

$$\rho c \partial_t T + \partial_x q = 0,$$

$$B - \frac{1}{T} = k \partial_x q$$

$$q = l_1 \partial_x B - \hat{l}_{12} \xi$$

$$\partial_t \xi = l_{21} \partial_x B - \hat{l}_2 \xi$$

General constitutive relation

$$\partial_t q + \hat{l}_2 q - L \partial_x \left(\frac{1}{T} + k \partial_x q \right) - l_1 \partial_x \partial_t \left(\frac{1}{T} + k \partial_x q \right) = 0$$

$$L = (l_1 \hat{l}_2 - \hat{l}_{12} l_{21})$$

General constitutive relation

$$\tau \partial_t q + q + \lambda_F \partial_x T - a_1 \partial_{xx} q - \lambda_1 \partial_{tx} \left(\frac{1}{T} \right) - a_2 \partial_{txx} q = 0$$

Guyer-Krumhansl hierarchy

Balance + constitutive, non-dimensional

$$\begin{aligned}\partial_t T + \partial_x q &= 0, \\ \tau \partial_t q + q + \lambda \partial_x T - a \partial_{xx} q &= 0\end{aligned}$$

Wave form

$$\tau \partial_{tt} T - \lambda_F \partial_{xx} T + \partial_t T - a \partial_{txx} T = 0$$

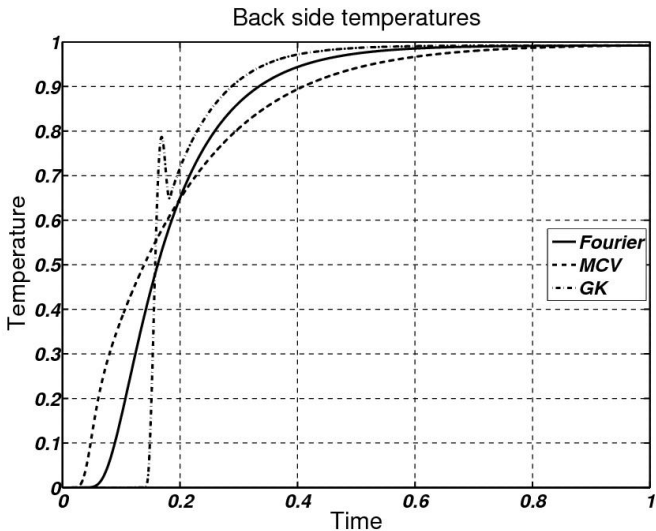
Hierarchy

$$\tau \partial_t \left(\partial_t T - \frac{a}{\tau} \partial_{xx} T \right) + \partial_t T - \lambda \partial_{xx} T = 0$$

Some solutions are exactly **Fourier!**

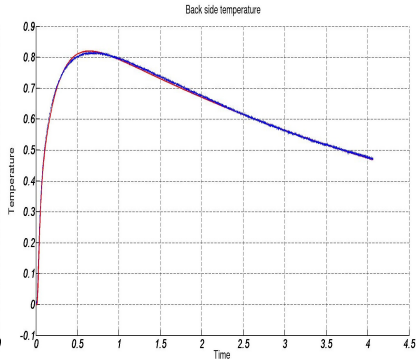
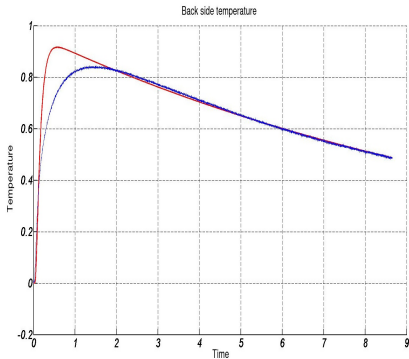
Experiments?


- Fourier, Maxwell-Cattaneo-Vernotte and Guyer-Krumhansl




Experiments?

- Metal foam: Fourier and Guyer-Krumhansl



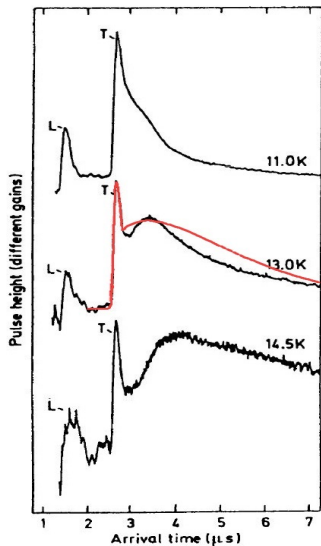
 V.P. and Fülöp T. (2012) Ann. Phys., 524(8), 470.

 Kovács R. and V.P. (2015), Int. J. Heat and Mass Transfer, 83, p613-620.

 Both, S. et al. (2016), J. Non-Equilibrium Thermodynamics, 41(1), p41-48.

Experiments?

Low temperature NaF experiments:
ballistic propagation and second sound
(courtesy of R. Kovács)



Rheology of solids

Elasticity with an internal variable I

Extended state space and consequences

$$\tilde{s}(e, \epsilon, \xi) = s(e, \epsilon) - \frac{a}{2}\xi^2$$

$$\rho ds = \frac{\rho}{T} de - \frac{\sigma}{T} d\epsilon - \rho a \xi d\xi$$

$$\tilde{\sigma}(e, \epsilon, \xi) = \sigma(e, \epsilon) + \hat{\sigma}(e, \epsilon, \xi)$$

$$\rho \partial_t e + \partial_x q = \tilde{\sigma} \partial_t \epsilon = \sigma \partial_t \epsilon + \hat{\sigma} \partial_t \epsilon$$

Entropy production

$$\rho \partial_t \tilde{s} + \partial_x j_s = \frac{\rho}{T} \partial_t e - \frac{\sigma}{T} \partial_t \epsilon - \rho a \xi \partial_t \xi + \partial_x \frac{q}{T} = q \partial_x \frac{1}{T} + \frac{\hat{\sigma}}{T} \partial_t \epsilon - \rho a \xi \partial_t \xi \geq 0$$

Mechanical part:

$$(\tilde{\sigma} - \underbrace{T \rho \partial_{\epsilon} s}_{E_{\epsilon}}) \partial_t \epsilon - A \xi \partial_t \xi = \hat{\sigma} \partial_t \epsilon - A \xi \partial_t \xi \geq 0$$

Elasticity with an internal variable II

Linear solution: beyond viscoelasticity

$$\hat{\sigma} = l_{11}\partial_t\epsilon + l_{12}(-A\xi), \quad \partial_t\xi = l_{21}\partial_t\epsilon + l_{22}(-A\xi)$$

Second law restrictions:

$$l_{11} \geq 0, \quad l_{22} \geq 0, \quad \det(L^s) \geq 0.$$

Elimination of the internal variable

$$\tilde{\sigma} + \frac{1}{Al_{22}}\partial_t\tilde{\sigma} = E\epsilon + \left(\frac{\det(L)}{l_{22}} + \frac{E}{Al_{22}}\right)\partial_t\epsilon + \frac{l_{11}}{Al_{22}}\partial_{tt}\epsilon$$

Kluitenberg-Verhás body

$$(\tilde{\sigma} - E\epsilon) + \tau\partial_t(\tilde{\sigma} - E_1\epsilon) = E_2\partial_{tt}\epsilon$$

Hooke, Kelvin-Voigt, Poynting-Thomson-Zener (standard) bodies.

Must be doubled in 3D isotropy!

Experiments?

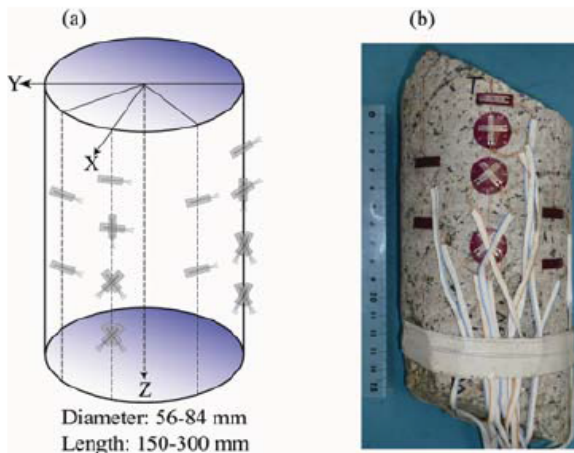


Rock 2





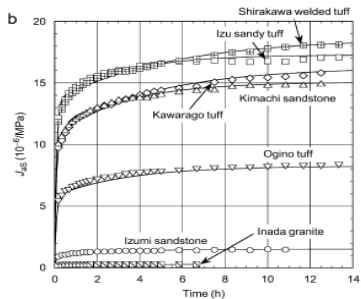
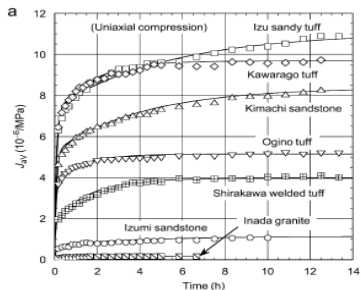
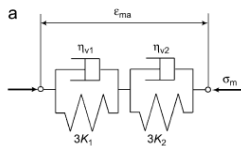
Anelastic Strain Recovery



Kluitenberg-Verhás

Deviatoric and volumetric relaxation

Matsuki and Takeuchi (1991-99)
Matsuki (2008), Lin et al. (2010)



Hierarchy in time:

- Second law: nonnegative coefficients
- Isotropy: separation of deviatoric and spherical parts, like Lamé coefficients or bulk and shear viscosities.
- Spherical and deviatoric Kluitenberg for uniaxial loading:

$$\begin{aligned}(\sigma - E\epsilon) + \tau\partial_t(\sigma - E_1\epsilon) + \tau_1\partial_{tt}(\sigma - E_2\epsilon) + \tau_2\partial_{ttt}(\sigma - E_3\epsilon) \\ = E_4\partial_{tttt}\epsilon\end{aligned}$$

- Material - apparent: deviatoric Kelvin-Voigt + spherical Hooke = Kluitenberg-Verhás in uniaxial compression.



Asszonyi Cs., Fülöp T. and V.P., Continuum Mechanics and Thermodynamics, (2015), 27, p971-986, (arXiv:1407.0882).

Summary - hierarchies galore

Ordinary thermodynamics

- Thermodynamics is stability
- Inertia leads mechanics

Wave approach to heat conduction

- Current multiplier
- Fourier equation hierarchy

Elasticity with a single internal variable of state

- Kluitenberg body
- Hooke law hierarchy

Elasticity with dual weakly nonlocal internal variables

- Modified entropy current
- Wave equation hierarchy

Thank you for the attention!

Ordinary thermodynamics:

- Truesdell, C. and Bharatha, S., Classical Thermodynamics as a Theory of Heat Engines, Springer, 1977.
- Matolcsi, T., Dynamical laws in thermodynamics, Physics Essays, 1992, 5/3, p320-327.
- Matolcsi, T., Ordinary Thermodynamics, Academic Publisher, Budapest, 2005.

Heat conduction:

- Both et al, Deviation from Fourier law at room temperature heterogeneous materials, Journal of Non-Equilibrium Thermodynamics, (2016), 41/1, 41-48, (arXiv:1506.05764).

Rheology:

- Asszonyi Cs., Fülöp T. and V.P., Distinguished rheological models for solids in the framework of a thermodynamical internal variable theory, Continuum Mechanics and Thermodynamics, (2015), 27, 971-986, (arXiv:1407.0882).